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PENTAFLUOROPHENYL DERIVATIVES OF CHLORINE TRIFLUORIDE

J. A. Obaleye*, R. Rahbarnooi and L. C. Sams

Texas Woman's University, Denton, Texas (U.S.A.) and Department of Chemistry, University of Calabar, Cross River State (Nigeria)

A reaction between C_6F_5Cl and elemental fluorine under pressure produced pentafluorophenylchlorine(III)difluoride⁽¹⁾ and 1,2-difluoro-1,2-bis(pentafluorophenyl)dichlorane, $C_6F_5Cl(F)Cl(F)C_6F_5$.

The dichlorane was a stable colorless liquid with a boiling point of 121 - 123°. Its characterization by elemental analysis, IR, NMR and Mass Spectra will be presented. It is believed to be the first stable compound with a chlorine-chlorine bond and may be considered to be a derivative of the unknown Cl_2F_A .

1 J.A. Obaleye and L.C. Sams, J. Fluorine Chem. 18 (1981) 31.

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SYNTHESIS AND PROPERTIES OF FLUORINATED ETHYL-METHYL-ETHERS

H. Muffler and G. Siegemund*

Hoechst AG, D-6230 Frankfurt 80 (F.R.G.)

Starting from trichloro-acetaldehyde the two ethers ${\rm CF_3}$ -CHCl-O-CH $_3$ and ${\rm CF_3}$ -CHF-O-CH $_3$ are available by three reaction steps including a catalytic gas-phase fluorination. These ethers are used as key compounds in the preparation of fluorinated ethers of the general formula

$$CF_3$$
-CHX-O-CH_{3-n}X_n X=F,C1; n=1-3

The conditions of the photochemical chlorination and the advantages of the different used fluorination methods concerning the chlorine-fluorine exchange in the chlorinated methyl groups will be presented. The physical, chemical and anaesthetic properties of the obtained derivatives will be mentioned.